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**Clinopyroxene-melt element partitioning during interaction between trachybasaltic magma  
and siliceous crust: Clues from quartzite enclaves at Mt. Etna volcano**

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**Abstract**

A peculiar characteristic of the paroxysmal sequence that occurred on March 16, 2013 at the New South East Crater of Mt. Etna volcano (eastern Sicily, Italy) was the eruption of siliceous crustal xenoliths representative of the sedimentary basement beneath the volcanic edifice. These xenoliths are quartzites that occur as subspherical bombs enclosed in a thin trachybasaltic lava envelope. At the quartzite-magma interface a reaction corona develops due to the interaction between the Etnean trachybasaltic magma and the partially melted quartzite. Three distinct domains are observed: (i) the trachybasaltic lava itself (Zone 1), including Al-rich clinopyroxene phenocrysts dispersed in a matrix glass, (ii) the hybrid melt (Zone 2), developing at the quartzite-magma interface and feeding the growth of newly-formed Al-poor clinopyroxenes, and (iii) the partially melted quartzite (Zone 3), producing abundant siliceous melt. These features makes it possible to quantify the effect of magma contamination by siliceous crust in terms of clinopyroxene-melt element partitioning. Major and trace element partition coefficients have been calculated using the compositions of clinopyroxene rims and glasses next to the crystal surface. Zone 1 and Zone 2 partition coefficients correspond to, respectively, the chemical analyses of Al-rich phenocrysts and matrix glasses, and the chemical analyses of newly-formed Al-poor crystals and hybrid glasses. For clinopyroxenes from both the hybrid layer and the lava flow expected relationships are observed between the partition coefficient, the valence of the element, and the ionic radius. However, with respect to Zone 1 partition coefficients, values of Zone 2 partition coefficients show a net decrease for transition metals (TE), high-field strength elements (HFSE) and rare earth elements including yttrium (REE+Y), and an increase for large ion lithophile elements (LILE). This variation is associated with coupled substitutions on the M1, M2 and T sites of the type  $^{M1}(\text{Al}, \text{Fe}^{3+}) + ^T\text{Al} = ^{M2}(\text{Mg}, \text{Fe}^{2+}) + ^T\text{Si}$ . The different incorporation of trace elements into clinopyroxenes of hybrid origin is controlled by cation substitution reactions reflecting local charge-balance requirements. According to the lattice strain theory, simultaneous cation exchanges across the M1, M2, and T sites have profound effects on REE+Y and HFSE partitioning. Conversely, both temperature and melt

composition have only a minor effect when the thermal path of magma is restricted to  $\sim 70$  °C and the value of non-bridging oxygens per tetrahedral cations (NBO/T) shifts moderately from 0.31 to 0.43. As a consequence, Zone 2 partition coefficients for REE+Y and HFSE diverge significantly from those derived for Zone 1, accounting for limited cation incorporation into the newly-formed clinopyroxenes at the quartzite-magma interface.

Keywords: clinopyroxene-melt element partitioning, magma contamination, quartzite xenoliths, Mt. Etna volcano.

## 1. Introduction

Mafic magmatism is generally invoked as the primary agent of crustal melting processes associated with a wide spectrum of contaminated magmas (Grove et al., 1988; Bergantz and Dawes, 1994; Mollo et al., 2011; Mollo and Vona, 2014). During early stages of contamination, rates of crustal assimilation can substantially exceed those of crystallization, with remarkable effects on trace element concentrations and isotopic evolution of magma (Huppert and Sparks, 1985; Reiners et al., 1995). Interpreting the geochemical compositions of mafic magmas depends on recognizing that their passage through a lithologically heterogeneous crust presents opportunities for assimilation of crustal materials via open-system processes. Under such circumstances, the degree of contamination is dictated by magma and crust compositions, temperature, incubation time, and rate of magma supply (Hildreth and Moorbath, 1988; Huppert and Sparks, 1988; Kerr et al., 1995; Cribb and Barton, 1996; Gamble et al., 1999; Dungan and Davidson, 2004; Erdmann et al., 2007).

The occurrence of xenocrysts and xenoliths in lavas indicate that assimilation of crustal lithologies is a dynamic process characterized by multiple stages and mechanisms of dissolution (i.e., mechanical disaggregation and partial melting) leading to formation of contaminated melts and refractory xenocrysts. Digestion of xenoliths in mafic magmas may be extraordinarily rapid with the potential to impose significant geochemical overprints on host magmas due to mixing with anatectic

melts (Costa and Dungan, 2005; Erdmann et al., 2007). Xenoliths in trachybasaltic lavas at Mt. Etna volcano (eastern Sicily, Italy) display mineralogical and geochemical changes associated with partial-to-complete melting of crustal materials that contribute to selective alkali enrichments in the enclosing magma (Michaud, 1995). An interesting peculiarity of siliceous xenoliths is the development of high-temperature reaction coronas at the xenolith-magma interface in which distinctive clinopyroxenes crystallize from differentiated alkaline melts of hybrid origin (Michaud and Clochiatti, 1995). This study attempts to quantify the geochemical effect of magma contamination by siliceous crust in terms of clinopyroxene-melt element partitioning. Although the discussion is limited in scope to the partial melting of quartzite xenoliths, the overall geochemical characteristics of natural products are relevant to other situations that involve mixing between SiO<sub>2</sub>-rich anatectic melts and trachybasaltic magmas. These contamination phenomena favor the crystallization of diopside-rich clinopyroxenes compositionally distinct from those resulting from closed-system differentiation of the host magma. Thermodynamic considerations based on the lattice strain theory show that trace element partitioning is controlled primarily by abrupt changes in clinopyroxene chemistry, whereas minor effects can be attributed to temperature and melt composition at the conditions observed in this study. Melt hybridization processes accompanied by clinopyroxene crystallization cause unexpected trace element partitioning with respect to changes observed during magma differentiation by either fractional crystallization or assimilation and fractional crystallization (AFC) mechanisms.

## 2. Geological background

Mt. Etna, the most active basaltic volcano in Europe, developed in a complex tectonic context at the intersection of tensional tectonic fractures that cut a 10-15 km-thick sedimentary-metamorphic pile, composed of two main structural units: the autochthonous carbonate unit of the Hyblean Plateau (the undeformed foreland domain of the African continental plate margin) and the overlying allochthonous flysch series of the Apenninic-Maghrebian Chain (Branca et al., 2011).

The onset of volcanism occurred with the eruption of tholeiitic basalts (ca. 500 ka) and, subsequently (ca. 220 ka), the composition of magmas shifted towards a more alkaline affinity (Tanguy et al., 1997). Volcanic activity is characterized by common summit eruptions and less frequent flank eruptions that are both controlled principally by rapid magma ascent and decompression along a vertically-developed central conduit (Corsaro et al., 2009). A distinctive feature of the recent activity is the relatively constant degree of magmatic differentiation with systematic eruption of trachybasaltic magmas containing a uniform phenocryst assemblage of olivine, clinopyroxene, plagioclase, and titanomagnetite (Ferlito et al., 2011; Armienti et al., 2013; Lanzafame et al., 2013; Mollo et al., 2013a; Giacomoni et al., 2014). The plumbing system is governed by frequent inputs from mantle depths of primitive, volatile-rich magmas that mix with more evolved, degassed melts residing at shallow crustal levels (Clocchiatti et al., 2004; Métrich et al., 2004; Spilliaert et al., 2006; Ferlito et al., 2008; Mollo et al., 2011a; Kahl et al., 2015).

During the January 2011 - April 2013 paroxysmal sequence at Mt. Etna, the New South East Crater of the volcano was characterized by several episodes of lava fountaining (Behncke et al., 2014; Viccaro et al., 2015). The trachybasaltic scoria clasts from these events show the typical phase assemblage of Etnean eruptions, with porphyritic index ranging between 15 and 30 vol.%. Both crystals and host magmas were rapidly quenched at the contact with the atmosphere, preserving the original compositions attained under intratelluric conditions (Mollo et al., 2015a). The crystallization path of magmas feeding the lava fountains has been estimated using the compositions of clinopyroxenes phenocrysts and matrix glass from scoria clasts as input data for thermometers, barometers and hygrometers (Mollo et al., 2015a; Perinelli et al., 2016). The estimates indicate that the saturation temperature of clinopyroxene decreases from ~1,150 to ~1,050 °C along a decompression path from ~800 to ~0.1 MPa in which the melt-H<sub>2</sub>O concentration decreases from ~4.5 to ~1.5 wt.%. Changes in the *P-T-H<sub>2</sub>O* crystallization conditions of clinopyroxene reflect the shift from H<sub>2</sub>O-undersaturated to H<sub>2</sub>O-saturated regimes (Armienti et al., 2013). Volatile-rich magmas stored in the deeper parts of the plumbing system are persistently

buffered to the composition of trachybasalt by low degrees of fractional crystallization in the order of 10-20 vol.% (Corsaro et al., 2013; Mollo et al., 2015b; Vetere et al., 2015). Conversely, at shallower crustal levels, Etnean magmas undergo strong degassing and crystallization while travelling from the conduit to the surface (Métrich et al., 2004; Spilliaert et al., 2006). On eruption, the phenocryst and microphenocryst content of magma is up to 50 vol.% but a great number of crystals nucleate and grow by degassing- and cooling-driven crystallization mechanisms (Applegarth et al., 2013).

A peculiar characteristic of the paroxysmal sequence on March 16, 2013 at Etna's New South East Crater was the eruption of siliceous crustal xenoliths representative of the sedimentary basement beneath the volcano. These xenoliths are quartzites from the stratigraphic formations (i.e., sands, sandstones, quartzites, and quartz arenites) of the Sicilide Tectonic Unit (~3 km depth, ~100 MPa lithostatic pressure) belonging to the Apenninic-Maghrebian Chain (Branca et al., 2011). Both crustal (Michaud, 1995) and cognate (Corsaro et al., 2014) xenoliths are commonly found in the prehistoric and historic Etnean eruptions as heralds of shallow-to-deep interaction processes between sedimentary, metamorphic, or igneous rocks and ascending trachybasaltic magmas. Additionally, changes in the physico-chemical properties of silicate and carbonate sedimentary rocks under the effect of thermally-induced reactions are addressed as source of (i) anomalously low seismic velocity zones at depth (Mollo et al., 2011), (ii) volcanic edifice instability mechanisms (Mollo et al., 2012a), (iii) extremely high H<sub>2</sub>O (silicate mineral dehydroxylation) and CO<sub>2</sub> (carbonate mineral decarbonation) emissions (Michaud, 1995; Mollo et al., 2011; Heap et al., 2013) and (iv) selective contamination of the magma with enrichments in K, Rb and Cs by percolation of a carrier fluid phase through the upper part of the magmatic feeding system (Michaud, 1995).

### 3. Analytical methods

Textural and microchemical analyses were carried out at the HP-HT Laboratory of Experimental Volcanology and Geophysics of the Istituto Nazionale di Geofisica e Vulcanologia in

Rome (Italy). Images were collected in backscattered electron (BSE) mode of a field emission gun-scanning electron microscopy (FE-SEM) Jeol 6500F equipped with an energy-dispersive spectrometer (EDS) detector. Microchemical analyses (Table 1S) were performed using an electron probe micro-analyzer (EPMA) Jeol-JXA8200 with combined EDS-WDS (five spectrometers with twelve crystals). Data were collected using 15 kV accelerating voltage and 10 nA beam current. For glasses, a slightly defocused electron beam with a size of 3  $\mu\text{m}$  was used, with a counting time of 5 s on background and 15 s on peak. For crystals, the beam size was 1  $\mu\text{m}$  with a counting time of 20 and 10 s on peaks and background respectively. The following standards were used: jadeite (Si and Na), corundum (Al), forsterite (Mg), andradite (Fe), rutile (Ti), orthoclase (K), apatite (P), and spessartine (Mn). Sodium and potassium were analyzed first to minimize alkali migration effects. The precision of the microprobe was measured through the analysis of well-characterized synthetic oxide and mineral secondary standards. Based on counting statistics, analytical uncertainties relative to their reported concentrations indicate that precision was better than 5% for all cations. BSE imaging combined with X-ray mapping were adopted for the identification of the main textural and compositional features of each sample. X-ray maps were collected for Si, Al, Fe, Mg, and Ca by using 15 kV accelerating voltage, 8 nA probe current, resolution  $1024 \times 768$  pixel<sup>2</sup>, and dwell time 10 ms (real time) per pixel. To the single-band images, representing X-ray maps for Si and Al, were assigned different colors (blue and red, respectively) and combined to form a colored three-band image. Phases containing a combination of more than one of the elements being mapped are displayed with composite colors.

Trace element analyses were conducted at the CNR-IGG-Pavia by LA-ICP-MS (Laser Ablation Inductively Coupled Plasma Mass Spectrometry). The chosen trace elements are representative of the following geochemical groups (Table 1S): rare earth elements (REE) comprising both light (LREE; La, Ce, Pr, Nd, Sm, Eu, and Gd) and heavy (HREE; Tb, Dy, Ho, Er, Tm, Yb and Lu, plus Y) elements, high field strength elements (HFSE; Ti, Hf, and Zr), large-ion lithophile elements (LILE; Sr, Pb, and Ba), and transition elements (TE; Cr and Sc). The laser



source used for the analyses consists of a Q-switched Nd:YAG laser (Brilliant, Quantel), with a fundamental emission in the near-IR region (1064 nm) which is converted into 266 nm by two harmonic generators. Using mirrors, the laser beam is carried into a petrographic microscope, focused above the sample, and then projected onto it. Optimum average instrumental operating conditions are: RF power 800-900 W, cooling gas 12.08 l min<sup>-1</sup>, sample gas 0.9-1.1 l min<sup>-1</sup>, auxiliary gas 1.00 l min<sup>-1</sup> and carrier gas 0.9-1.1 l min<sup>-1</sup>. The total scan-time is about 700 ms, the settling time is about 340 ms, and hence the acquisition efficiency is estimated at about 50%. A typical analysis consists of acquiring one minute of background and one minute of ablated sample, thus approximately 170 sweeps are required. The mean integrated time for acquisition is about 0.9 s for each element. A 10 µm spot size was used for LA-ICP-MS measurements. The ablated material was analyzed with a quadrupole (DRCe, PerkinElmer). NIST-SRM610 was used as an external standard; <sup>43</sup>Ca or <sup>29</sup>Si were adopted as internal standards for clinopyroxene and glass analyses respectively. In each analytical run the USGS reference sample BCR2 was analyzed together with the unknowns for quality control. The precision of individual analyses varied depending upon a number of factors, e.g., the element and isotope analyzed as well as the chemical homogeneity of crystal and glass. However, the 1 sigma errors calculated from variations in replicate analyses of crystals and glasses were invariably several times larger than the fully integrated 1 sigma errors determined from counting statistics alone. Trace element Nernst partition coefficients [ $D_i = (I)_{\text{xls}} / (I)_{\text{melt}}$  on a weight basis] were calculated using the clinopyroxene rim and coexisting glass analyses (Table 2S). To avoid contamination caused by the partial ablation of crystals, the glass was analyzed 3 µm away from the clinopyroxene edge. For the case of the trachybasaltic lava, partition coefficients (i.e., Zone 1 partition coefficients) were derived using the compositions of clinopyroxene phenocrysts and matrix glasses, as representative of the quenched melt phase supplying chemical components to the growing crystals. Conversely, for the case of the quartzite-magma interface, partition coefficients (i.e., Zone 2 partition coefficients) were derived using the

compositions of newly-formed clinopyroxenes and coexisting glasses, as representative of the hybrid melt from which crystals nucleated and grew.

Bulk rock analyses for major and trace elements, as well as oxygen isotope analyses were conducted at Actlabs (Activation Laboratories Ltd.). These analyses are reported in the supplementary material (Table 3S), together with measured and certified analyses of international reference materials. Major elements were analyzed by lithium metaborate/tetraborate fusion – ICP-OES (inductively coupled plasma optical emission spectrometry). Sample washing was performed to remove any organic, loosely adhered, and cementing material. Washing included soaking in a hot mixture of HCl and H<sub>2</sub>O<sub>2</sub> as well as cleaning in acetone using ultrasound. The analyses were performed in a batch system. Each batch contained a method reagent blank, certified reference material and 17% replicates. Samples were mixed with a flux of lithium metaborate and lithium tetraborate and fused in an induction furnace. The melted material was immediately poured into a solution of 5% nitric acid containing an internal standard, and mixed continuously until completely dissolved (about 30 minutes). The samples were run on a Thermo Jarrell-Ash ENVIRO II ICP. Calibration was performed using 7 prepared USGS and CANMET certified reference materials. FeO was determined through titration, using a cold acid digestion of ammonium metavanadate, and hydrofluoric acid in an open system. Ferrous ammonium sulphate was added after digestion and potassium dichromate was the titrating agent. This cold digestion dissolved silicates and some sulphides. Water content was determined gravimetrically. About 0.3 g sample was thermally decomposed in a resistance furnace in a pure nitrogen environment at 110 °C (moisture, H<sub>2</sub>O<sup>-</sup>) followed by decomposition at 1,000 °C (interstitial water, H<sub>2</sub>O<sup>+</sup>), using an ELTRA CW-800, directly releasing H<sub>2</sub>O. Trace elements were measured by lithium metaborate/tetraborate fusion – ICP-MS (inductively coupled plasma mass spectrometry). Samples fused as reported above were diluted and analyzed by Perkin Elmer Sciex ELAN 6000, 6100 or 9000 ICP/MS. Three blanks and five controls (three before sample group and two after) were analyzed per group of samples. Oxygen isotopes were measured by reaction with BrF<sub>5</sub> at about 650 °C in nickel bombs. The

fluorination reaction converts O in the minerals to O<sub>2</sub> gas, which is subsequently converted to CO<sub>2</sub> gas using a hot C rod. All reaction steps are quantitative. Isotopic analyses were performed on a Finnigan MAT Delta, dual inlet, isotope ratio mass spectrometer. Oxygen isotope ratios are reported in the standard delta notation as per mil deviations from V-SMOW (hereafter reported to as SMOW). External reproducibility is  $\pm 0.19\%$  (1 sigma error).

## 4. Results

### 4.1 Textural characters

Quartzites of this study occur as subspherical bombs with maximum length of 4-15 cm enclosed in a thin trachybasaltic lava envelope (Fig. 1a; cf. Michaud, 1995). The interaction between the trachybasaltic magma and the quartzite can be represented by three distinct domains: (i) trachybasaltic lava (Zone 1; Fig. 1b,c,f), (ii) hybrid melt forming a variably thick (~50-500  $\mu\text{m}$ ) boundary layer between lava and quartzite (Zone 2; Fig. 1b,c,d,e), and (iii) partially melted quartzite (Zone 3; Fig. 1b,c,d,e). Zone 1 displays the typical phase assemblage found in recent trachybasaltic Etnean eruptions (cf. Mollo et al., 2015b) that, in order of abundance, comprises tabular plagioclase, prismatic clinopyroxene, subrounded olivine and titanomagnetite dispersed in a vesicular matrix glass (Fig. 1b,c,f). Zone 2 is characterized by melt hybridism associated with crystallization of clinopyroxenes chemically different from those in host lava (Fig. 1b,c,d,e). The texture of Zone 2 clinopyroxenes changes from euhedral to anhedral with maximum size (~50  $\mu\text{m}$ ) smaller than that (~120  $\mu\text{m}$ ) generally observed for Zone 1 prismatic crystals (Fig. 1f). Rarely, the hybrid melt may be thick enough so that larger (~100  $\mu\text{m}$ ) clinopyroxenes also occur in Zone 2, exhibiting well-formed planar edges (Fig. 1e). These texturally mature minerals may form by aggregation of smaller and mutually touching smaller crystals, resembling crystal growth mechanism documented by ex-situ (Pupier et al., 2008; Iezzi et al., 2011, 2014; Mollo et al., 2012b) and in-situ (Schiavi et al., 2009) experiments. Zone 3 is made of subrounded quartz crystal relics embedded in a vesicular interstitial glass. The size of quartz grains changes significantly from ~20

to ~700  $\mu\text{m}$  (Fig. 1b,c,d,e) but most of the smaller crystals occur in close contact with the trachybasaltic host at the periphery of the subspherical bomb (Fig. 1d) where the degree of partial melting is maximum (cf. Michaud, 1995).

#### 4.2 Clinopyroxene chemistry and crystallization

The comparison between clinopyroxenes from Zone 1 and Zone 2 reveals that crystals from the trachybasaltic lava show  $^{\text{T}}\text{Al}/\text{Si}$  ratios (~0.1-0.13; Al and Si are expressed as atoms per formula unit) much higher than those (~0.02-0.04) measured in the hybrid melt (Fig. 2). Contrary to what is observed in Zone 2, clinopyroxenes from Zone 1 are also enriched in Ti,  $\text{Fe}_{\text{tot}}$ , Na, and Cr, counterbalanced by depletions in Mg and Ca (Fig. 2). The charge balance equation of Lindsley (1980) for the nominal determination of  $\text{Fe}^{3+}$  indicates that the  $\text{Fe}^{2+}/\text{Fe}^{3+}$  ratio of clinopyroxene increases from ~2 to ~6 from Zone 1 to Zone 2. Accordingly, clinopyroxenes from the hybrid melt are characterized by cation exchanges in which  $^{\text{M1}}(\text{Al}, \text{Fe}^{3+})$  substitute for  $^{\text{M2}}(\text{Mg}, \text{Fe}^{2+})$  coupled with replacement of Al with Si in the tetrahedral site. As a consequence, hedenbergite (Hd), enstatite (En), ferrosilite (Fs), jadeite (Jd) and tschermakitic components ( $\Sigma\text{T}$ s equals to the sum of Ca-, CaFe-, CaCr- and CaTi-Tschermak molecules) are preferentially incorporated in clinopyroxenes from the trachybasaltic lava (Fig. 2), whereas clinopyroxenes from the hybrid melt are enriched in diopside (Di). Despite these compositional variations, clinopyroxenes from Zone 1 and Zone 2 are invariably classified as diopside by the scheme of Morimoto (1988). The crystallization of Ti-Al-poor and Si-rich diopsidic crystals has been previously documented in the reaction coronas found at the interface between quartzite xenoliths and trachybasaltic lavas erupted during historic volcanic activity of Mt. Etna (Michaud and Clocchiatti, 1995).

To test whether clinopyroxenes from Zone 1 and Zone 2 were effectively in equilibrium with melts supplying chemical components to the growing crystals, the equilibrium model of Mollo et al. (2013b) was used. Clinopyroxene-melt pairs used for calculations are those reported in Table 1S, corresponding to the analyses of clinopyroxenes and glasses used for the calculation of Zone 1

and Zone 2 partition coefficients. The equilibrium is based on the difference ( $\Delta$ ) between diopside + hedenbergite (DiHd) contents measured in the analyzed crystals with those predicted for clinopyroxene via regression analysis of equilibrium clinopyroxene-melts pairs. Considering that  $\Delta$ DiHd should equal to zero at thermodynamic equilibrium, the derived values (0.01–0.08) suggest near-equilibrium crystallization (Fig. 3a), plotting within 10% of the one-to-one line (cf. Mollo and Masotta, 2013). As a further test, we have used the equilibrium model of Putirka (2008) based on the Fe–Mg cation exchange reaction [ $K_{\text{Fe-Mg}} = (\text{Fe}^{\text{cpx}} / \text{Fe}^{\text{melt}}) \times (\text{Mg}^{\text{melt}} / \text{Mg}^{\text{cpx}})$ ]. The measured values for  $K_{\text{Fe-Mg}}$  (0.24–0.27; Fig. 3b) closely match with the equilibrium ranges of  $0.27 \pm 0.03$  and  $0.28 \pm 0.08$  indicated by Putirka et al. (2003) and Putirka (2008) respectively. These equilibrium clinopyroxene compositions were also used as input data for the  $P$ - $\text{H}_2\text{O}$ -independent thermometer of Putirka et al. (1996) with uncertainty  $\pm 23$  °C. Results show that Zone 1 clinopyroxenes formed from the trachybasaltic lava within a thermal range ( $\sim 1,140$ – $1,170$  °C) consistent with the early saturation temperature of clinopyroxenes in Etnean magmas (Mollo et al., 2015b). Conversely, Zone 2 clinopyroxenes from the hybrid melt equilibrated at lower temperature conditions ( $\sim 1,070$ – $1,130$  °C), suggesting crystallization when the ascending trachybasaltic magma encountered and interacted with the colder, shallower crustal basement.

#### 4.3 Glass chemistry and melt hybridism

The residual melt from Zone 1 is chemically homogeneous basaltic trachyandesite (Fig. 4) with  $\text{SiO}_2 \approx 51$ – $52$  wt.% and  $\text{Na}_2\text{O} + \text{K}_2\text{O} \approx 7$ – $8$  wt.%. Hundreds of microns from the interaction Zone 2, the interstitial glass in close contact with the quartz grains from Zone 3 is significantly enriched in  $\text{SiO}_2$  (80–86 wt.%; Fig. 4), resembling the analyses of siliceous melts obtained by partial quartz dissolution (Watson, 1982). Conversely, the hybrid melt at the quartzite-lava interface from Zone 2 exhibits trachyandesitic to trachytic compositions ( $\text{SiO}_2 \approx 56$ – $68$  wt.% and  $\text{Na}_2\text{O} + \text{K}_2\text{O} \approx 8$ – $9$  wt.%; Le Maitre et al., 2002) intermediate between those from Zone 1 and Zone 3 (Fig. 4). Major oxide analyses for  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{FeO}$ ,  $\text{MgO}$ , and  $\text{CaO}$  almost plot on tie-lines between basaltic

trachyandesite and rhyolite end-members, indicative of a mixing process (Fig. 4). The contamination of the Etnean magma by progressive assimilation of a siliceous component is also evidenced by quantitative X-ray maps of the quartzite-lava interface (Fig. 5). The degree of hybridism between melts of markedly contrasting origin is better illustrated by Si-Al X-ray mapping overlays in which, among all major cations, silicon and aluminium show the greatest chemical variations (Fig. 6). From Zone 1 to Zone 3 via the hybrid melt of Zone 2, SiO<sub>2</sub> increases by ~69% relative, whereas Al<sub>2</sub>O<sub>3</sub> decreases by ~75%. With respect to the original Al<sub>2</sub>O<sub>3</sub>/SiO<sub>2</sub> ratio (~0.31) of the Etnean magma (Table 3S), the final Al<sub>2</sub>O<sub>3</sub>/SiO<sub>2</sub> ratio (~0.15) of the hybrid melt decreases by ~50%, resembling the <sup>T</sup>Al-Si cation exchange observed for clinopyroxene (Fig. 2). Notably, the crystallization of clinopyroxene from the hybrid melt causes deviations from pure mixing trends. Depletions in MgO and CaO are observed in the hybrid melt next to the advancing crystal surface, whereas enrichments in Na<sub>2</sub>O, K<sub>2</sub>O, SrO, and BaO reflect the highly incompatible character of these elements within the clinopyroxene crystal lattice (Fig. 4). To estimate the degree of clinopyroxene crystallization from the hybrid melt, linear least-squares mass balance calculations are performed following the same strategy adopted by Barnes et al. (2002). Specifically, the simple mixing process between the basaltic trachyandesite from Zone 1 and siliceous component from Zone 3 (i.e., mixing end-members) is coupled with the crystallization of newly-formed clinopyroxenes from Zone 2. The trachyandesitic to trachytic compositions of the hybrid melt are fairly reproduced when the proportions of siliceous component and newly-formed clinopyroxenes are in the ranges of 8-30% and 6-18% respectively (Fig. 7). The goodness of fits is corroborated by the low sum of squares of residuals ( $\sum r^2 \leq 0.23$ ).

#### 4.4 Bulk-rock geochemistry

Bulk-rock compositions for the trachybasaltic lava (Table 3S) are very similar to those that typify recent eruptions. The Na<sub>2</sub>O/K<sub>2</sub>O ratio (~1.8) indicates that the lava can be defined as K-trachybasalt, in agreement with the shift from Na- to K-affinity of products erupted after the 1971

event (Viccaro and Cristofolini, 2008). Trace element concentrations show common features for historical Etnean basalts with negative anomalies in HFSE that reflect the imprint of slab-derived fluids in arc magmas originating from an OIB-type mantle source with relative enrichments in LILE and LREE (Corsaro and Metrich, 2016). The oxygen isotope ratio (6.1‰  $\delta^{18}\text{O}$ ; Table 3S) is also consistent with bulk-rock data (6.1-7.1‰  $\delta^{18}\text{O}$ ) obtained for previous lava flows (Viccaro and Cristofolini, 2008). Conversely, the quartzite bulk-rock composition (Table 3S) is similar to those previously reported for siliceous crustal xenoliths ( $\text{SiO}_2 > 90$  wt.%) from 1982 to 1989 Etnean eruptions (Michaud, 1995) and from other volcanic settings (Harris and Chaumba, 2001; Fourie and Harris; 2011). The oxygen isotope composition of quartzite (12‰  $\delta^{18}\text{O}$ ; Table 3S) is higher than that measured for single quartz grains from siliceous crustal rocks (~8‰  $\delta^{18}\text{O}$ ), indicative of hydrothermal alteration phenomena (Harris and Chaumba, 2001; Fourie and Harris; 2011).

## 5. Discussion

### 5.1 Clinopyroxene-melt element partitioning

Zone 1 and Zone 2 partition coefficients discussed in this study are presented as Onuma diagrams (Onuma et al., 1968) in which elements are grouped by ionic charge for M1 and M2 clinopyroxene crystallographic sites and arranged from smallest to largest ionic radius (Shannon, 1976) within each group (Fig. 8). The graphical representation shows how strongly the partition coefficient is controlled by the properties of individual crystallographic sites. This excludes possible glass contamination effects during chemical analyses that, in turn, would yield irregular relationships among the partition coefficient, the valence of the element, and the ionic radius (Kennedy et al., 1993). As stated above, Zone 1 partition coefficients were derived using the compositions of clinopyroxene phenocrysts and those of the matrix glasses from the trachybasaltic lava. Conversely, Zone 2 partition coefficients were derived using the compositions of newly-formed clinopyroxenes and those of the hybrid glasses measured at the quartzite-magma interface. Both Zone 1 and Zone 2 partition coefficients of isoivalent cations plotted against ionic radius lie on

parabola-like curves. However, Zone 2 partition coefficients show  $D_{TE}$ ,  $D_{HFSE}$ , and  $D_{REE+Y}$  values lower than those measured for Zone 1 partition coefficients (Fig. 8). The opposite occurs for  $D_{LILE}$ , these divalent cations more favorably incorporated in newly-formed clinopyroxenes from Zone 2. Notably, Zone 1 partition coefficients closely match values derived for Etnean trachybasaltic magmas crystallized in the laboratory (Mollo et al., 2013c) and under natural conditions (Scarlato et al., 2014). Conversely, Zone 2 partition coefficients resemble those measured for more differentiated magmas with either alkaline or calcalkaline affinity, i.e. trachytes (Mahood and Stimac, 1990; Huang et al., 2006; Pappalardo et al., 2008), basaltic andesites (Zajacz and Halter, 2007), andesites (Luhr and Carmichael, 1980; Ewart and Griffin, 1994; Kleinn et al., 2000), and dacites (Bacon and Druitt, 1988).

Clinopyroxenes from both Zone 1 and Zone 2 show that LILE in M2 shift from mildly incompatible (e.g.,  $D_{Sr} < 0.9$ ) to highly incompatible (e.g.,  $D_{Ba} < 0.0009$ ; Fig. 8). REE+Y are incompatible within the crystal lattice but, from smallest to largest ionic radius, HREE (e.g.,  $D_{Yb} < 0.4$ ) are more easily accommodated in clinopyroxene relative to LREE (e.g.,  $D_{La} < 0.1$ ). Tetravalent cations from HFSE group are moderately incompatible in M1, showing  $D_{Ti} \approx D_{Hf} < 0.7$  and  $D_{Zr} < 0.2$  (Fig. 8). TE partition coefficients are either compatible ( $D_{Cr}$  and  $D_{Sc} > 1.7$ ) or incompatible ( $D_{Cr}$  and  $D_{Sc} < 1$ ) within clinopyroxenes from Zone 1 and Zone 2 respectively. Trace element partitioning of different valence cations onto M2 exhibits decoupled behavior characterized by increments of 2+ LILE cations (e.g., Sr) and decrements of 4+ HFSE cations (e.g., Hf). From Zone 1 to Zone 2 clinopyroxenes, the partitioning parabola shifts upward for divalent LILE cations on M2 (Fig. 8), whereas the opposite occurs for  $D_{TE}$ ,  $D_{HFSE}$ , and  $D_{REE+Y}$ , paralleling the decreasing number of Al cations in the tetrahedral site (Lindstrom, 1976; Ray et al., 1983; Hart and Dunn, 1993; Forsythe et al., 1994; Lundstrom et al., 1994, 1998; Skulski et al., 1994; Blundy et al., 1998; Hill et al., 2000; Wood and Trigila, 2001; Bedard, 2014). The different incorporation of trace elements in clinopyroxenes from Zone 1 and Zone 2 is related to charge balance mechanisms (Blundy and Wood, 1994, 2003; Wood and Blundy, 1997; Lundstrom et al., 1998) primarily



controlled by  ${}^T\text{Al}$  (Fig. 9). The entry of HFSE onto M1 in Zone 1 clinopyroxene is facilitated because the average charge on M1 increases with increasing  $\Sigma\text{T}$ s (Fig. 2) due to replacement of  $\text{Mg}^{2+}$  by  $\text{Fe}^{3+}$  and  $\text{Al}^{3+}$  (Wood and Trigila, 2001; Marks et al., 2004). Similarly, the dependence of  $D_{\text{REE+Y}}$  on  ${}^T\text{Al}$  is interpreted as the increasing probability of achieving charge-neutral local configurations (e.g.,  $\text{REEMgSiAlO}_6$ ) where REE+Y enter in a locally charge-balanced M2 site with increasing  ${}^T\text{Al}$  (Blundy et al., 1998). Such configurations do not involve electrostatic work of substitution of the trace element (Wood and Blundy, 2001) and, therefore, the Onuma parabola shifts upward in Al-rich clinopyroxenes from Zone 1 (Fig. 8). Conversely, decreasing Al concentrations in clinopyroxenes from Zone 2 make local charge balancing on M2 difficult, such that Al-poor clinopyroxenes cannot easily accommodate REE+Y into their crystal lattices (Hill et al., 2000; Wood and Trigila, 2001). It has been widely documented that REE+Y replace Ca in M2 while charge balanced by Al replacing Si in tetrahedral site or Na replacing Ca in M2 (Gaetani and Grove, 1995; Blundy et al., 1998; Schosnig and Hoffer, 1998; Bennett et al., 2004; Marks et al., 2004; Francis and Minarik, 2008; Sun and Liang, 2012). This compensation mechanism applies well to Zone 1 clinopyroxenes (Fig. 2). In contrast, Zone 2 crystals are significantly depleted in  ${}^T\text{Al}$  and Na greatly reducing the number of possible local charge-balanced configurations (cf. Wood and Blundy 2001). As a consequence, trivalent TE ( $\text{Sc}^{3+}$ ,  $\text{Cr}^{3+}$ ) decrease from Zone 1 to Zone 2 crystals (Fig. 9), due to a decreased net charge in M1 (Bedard, 2014) resulting from lower  $\text{Fe}^{3+}$  contents (Hill et al. 2000), whereas divalent LILE cations ( $\text{Sr}^{2+}$ ,  $\text{Pb}^{2+}$ ,  $\text{Ba}^{2+}$ ) increase in the M2 site of Zone 2 clinopyroxenes (Fig. 9), due to the higher  ${}^{\text{M2}}\text{Ca}$  in the more diopsidic crystals (cf. Michaud, 1995; Michaud and Clocchiatti, 1995).

## 5.2 Modeling trace element partitioning through the lattice strain theory

In Onuma diagrams, the parabolic trend of partition coefficients measured for an isoivalent series of cations can be described quantitatively by the lattice strain model (Brice, 1975; Blundy and Wood, 1994; Wood and Blundy, 1997):

$$D_i = D_0 \exp \left( \frac{-4\pi EN_A \left( \frac{r_0}{2} (r_i - r_0)^2 + \frac{1}{3} (r_i - r_0)^3 \right)}{RT} \right) \quad (1)$$

where  $D_i$  and  $r_i$  are, respectively, the partition coefficient and ionic radius of the element  $i$  of interest,  $D_0$  is the partition coefficient for the strain-free substitution,  $r_0$  is the radius of a hypothetical cation that substitutes into the site with zero strain,  $E$  is the effective Young's modulus for the lattice site,  $N_A$  is Avogadro's number ( $6.022 \times 10^{23} \text{ mol}^{-1}$ ),  $R$  is the universal gas constant ( $8.3145 \text{ J mol}^{-1} \text{ K}^{-1}$ ), and  $T$  is the temperature (in Kelvin). To obtain reliable estimates and minimize the standard error, the lattice strain model requires the determination of partition coefficients for a large range of isovalent cations (Yao et al., 2012; Sun and Liang, 2013; Bedard, 2014). On this basis,  $r_0$ ,  $D_0$ , and  $E$  have been determined for the partitioning of trivalent REE+Y, as these form the largest group of isovalent elements in the analytical dataset (Table 4S). Linear regression fits and their correlation coefficient ( $R^2$ ) show a high dependence of  $r_0$  ( $R^2 = 0.94$ ),  $D_0$  ( $R^2 = 0.95$ ), and  $E$  ( $R^2 = 0.94$ ) on  $^{\text{T}}\text{Al}$  of Zone 1 and Zone 2 clinopyroxenes (Fig. 10a).  $D_0$  increases due to the increasing probability of cations entering a locally charge-balanced site as the  $^{\text{T}}\text{Al}$  increases (Wood and Blundy 2001). Conversely,  $r_0$  and  $E$  decrease with increasing  $^{\text{T}}\text{Al}$  (Fig. 10a), corroborating the inverse correlation between  $r_0$  and  $^{\text{T}}\text{Al}$  (Adam and Green, 2006; Tuff and Gibson, 2007), as well as the positive linear dependence of  $E$  on  $r_0$  found for clinopyroxenes (Sun and Liang, 2012) and low-Ca pyroxenes (Yao et al., 2012).

If the melt composition has minor effects on  $D_{\text{REE+Y}}$ , the height of the partitioning parabola is expected to increase with increasing aluminum in the tetrahedral site and with decreasing temperature (Hill et al., 2000; Wood and Triguila 2001). However, the control of melt composition and structure on element partitioning cannot be excluded because the clinopyroxenes and melts result from a progressive mixing process between two compositionally distinct aluminosilicate liquids (Figs. 4, 5, and 6). Therefore, with respect to the uncontaminated trachybasaltic magma, the effect of  $^{\text{T}}\text{Al}$  on the lattice strain parameters can be mediated by the increasingly silicic character of

the hybrid melt (Gaetani 2004). Data from mafic melts show diffuse increase of  $D_{\text{REE+Y}}$  with  $^{\text{T}}\text{Al}$ , but data from felsic melts show less robust trends due to the influence of melt composition (Bedard, 2014). For clinopyroxenes with similar  $^{\text{T}}\text{Al}$  contents,  $D_{\text{REE+Y}}$  can increase by up to an order of magnitude as a result of the dominant effect of melt structure (Qian et al., 2015). The degree of melt polymerization may affect the trace element partitioning when the number of non-bridging oxygens per tetrahedral cations (NBO/T) is lower than 0.49 (Gaetani 2004). However, over a broad range of melt compositions the influence of melt structure on  $D_{\text{REE+Y}}$  is more effective below the NBO/T threshold of 0.25 (Huang et al., 2006; Mollo et al., 2015). NBO/T values from this study range from 0.31 to 0.43 and, therefore,  $D_0$  ( $R^2 = 84$ ) is poorly correlated with the degree of melt polymerization (Fig. 10b). Notably, the effect of melt structure is usually insignificant on  $r_0$  ( $R^2 = 81$ ) and  $E$  ( $R^2 = 81$ ) values (cf. Qian et al., 2015). On the one hand, it has been found that NBO/T is not a good predictor for the lattice strain model (Sun and Liang; 2012; Mollo et al., 2016). On the other hand, due to the similarity in ionic radius and charge, REE+Y are more likely to substitute for Ca in the melt relative to other divalent (Fe + Mg) and monovalent (Na + K) cations (Huang et al., 2006). When the influence of clinopyroxene chemistry is minimized,  $D_0$  (at fixed temperature) strictly depends on the melt structure expressed as the ratio of molar Ca / (Na + K + Fe + Ca + Mg) where cations are given as percentages (Huang et al., 2006). Data from this study exhibit a weak correlation ( $R^2 = 86$ ) between  $D_0$  and Ca / (Na + K + Fe + Ca + Mg) of the melt, testifying to rival effects between the increase in the number of large structural sites critically important to accommodating trace element cations in the melt and the strong control of clinopyroxene chemistry on REE+Y partitioning (Mollo et al., 2016). It is noteworthy that a regular solution model for predicting the effect of melt structure on REE+Y partition coefficient has been derived only when the melt composition changes drastically from gabbroic to granitic (Schmidt et al., 2006).

In order to better define the effect of clinopyroxene composition on REE+Y (i.e.,  $D_{\text{La}}$  and  $D_{\text{Lu}}$ ) and HFSE (i.e.,  $D_{\text{Ti}}$  and  $D_{\text{Zr}}$ ) partitioning, the dataset from this study has been compared with values predicted by the models of Wood and Blundy (1997) and Hill et al. (2011) (Table 5S). These

thermodynamically-derived equations are based on the lattice strain parameters and allow for discrimination and quantification of the primary compositional effects on trace element partitioning. Wood and Blundy (1997) parameterized the predictive model for REE+Y as functions of pressure ( $P$ ), temperature ( $T$ ),  $Mg$ -number of the melt [ $Mg\# = \text{molar Mg}/(\text{Mg}+\text{Fe})$ ] and  $^{M1}\text{Mg}$  of clinopyroxene, also considering that  $^{M2}\text{Ca}$  and  $^{M1}\text{Al}$  affect REE+Y partitioning through their influence on  $r_0$ . The input values used are the entrapment pressure (100 MPa) of the quartzite xenoliths, the crystallization temperature of clinopyroxenes (1,140-1,170 °C and 1,070-1,130 °C for Zone 1 and Zone 2, respectively), and major element compositions of crystal and coexisting glass (Table 1S). The control of pressure on  $D_{\text{REE+Y}}$  and  $D_{\text{Na}}$  is practically negligible at shallow crustal depths (Blundy et al., 1995; Blundy and Wood, 2003; Bennett et al., 2004). Additionally, the low (< 2 wt.%) concentration of  $\text{H}_2\text{O}$  dissolved in the Etnean magmas residing at shallow pressure (Perinelli et al., 2016) has negligible effects on the activity of  $\text{REEMgAlSiO}_6$  (Wood and Blundy, 2002).

$D_{\text{La}}$  and  $D_{\text{Lu}}$  are well predicted by the lattice strain model (Fig. 11), yielding high correlation coefficients ( $R^2_{\text{La}} = 0.96$  and  $R^2_{\text{Lu}} = 0.94$ ) and low standard errors of estimate ( $\text{SEE}_{\text{La}} = 0.002$  and  $\text{SEE}_{\text{Lu}} = 0.01$ ). As  $D_{\text{La}}$  and  $D_{\text{Lu}}$  decrease from Zone 1 to Zone 2 clinopyroxenes (Fig. 11), the shape of REE+Y partitioning remains subparallel in the Onuma diagram (Fig. 8), suggesting preferential control of crystal chemistry rather than melt structure (Hill et al., 2000; Wood and Trigila 2001; Qian et al., 2015). Considering the intrinsic calibration error of the model, values of  $r_0$  are also fairly reproduced by the expression ( $R^2 = 0.93$  and  $\text{SEE} = 0.001$ ) of Wood and Blundy (1997) based on  $^{M1}\text{Al}$  and  $^{M2}\text{Ca}$ , as well as by the more recent equation ( $R^2 = 0.94$  and  $\text{SEE} = 0.001$ ) of Sun and Liang (2012) based on  $^{M1}\text{Al}$  and  $^{M2}\text{Mg}$ . The shape of the partitioning parabola is determined by compositional variations from Zone 1 to Zone 2 clinopyroxenes in which lower  $^{\text{T}}\text{Al}$  is accompanied by decreasing  $^{M1}\text{Al}$  and increasing  $^{M2}\text{Mg}$  and  $^{M2}\text{Ca}$  during formation of Di-rich crystals (Fig. 2). Wood and Blundy (1997) modelled  $D_{\text{REE+Y}}$  as independent of  $^{\text{T}}\text{Al}$ , assuming simple short-range order between REE+Y on M2 site and Al in T site in the  $\text{REEMgAlSiO}_6$  molecule. Wood and

Trigila (2001) attempted to refit the lattice strain model by including  $^T\text{Al}$  but no improvements were found. Additionally, the prediction ability of the model does not change by including the electrostatic work for local charge balance between REE+Y in M2 and adjacent  $^T\text{Al}$  (Mollo et al., 2016). Looking at the test dataset used by Wood and Trigila (2001), it is apparent that values of  $^T\text{Al}$  and  $^{M1}\text{Al}$  decrease in concert with higher  $^{M1}\text{Mg}$  and  $^{M2}\text{Mg}$  contents, where  $^{M1}\text{Mg}$  is the expression of the activity of REEMgAlSiO<sub>6</sub> clinopyroxene ( $a_{\text{REEMgAlSiO}_6}^{\text{cpx}} = X_{\text{REE}}^{M2} X_{\text{Mg}}^{M1}$  in mole fraction) and  $^T\text{Al}$  is negatively correlated with  $^{M2}\text{Mg}$  (Sun and Liang, 2012). This leads to the conclusion that simultaneous cation exchanges across the M1, M2 and T sites, such as those observed in this study, make the  $^T\text{Al}$  term statistically unimportant for the original model of Wood and Blundy (1997). In other words, over a broad calibration dataset the dependence of  $D_{\text{REE+Y}}$  on  $^T\text{Al}$  is obscured by other compositional effects (cf. Wood and Trigila, 2001), such as those related to  $^{M2}\text{Ca}$  (Wood and Blundy, 2001),  $^{M2}\text{Mg}$  (Sun and Liang, 2012), and  $^{M1}\text{Fe}$  (Dygert et al., 2014).

For the case of HFSE, Hill et al. (2011) modelled  $D_{\text{Ti}}$  through a thermodynamic description for the crystal-melt exchange reaction and the energetics of the different charge-imbalanced configurations produced by insertion of a heterovalent cation. The authors derived expressions for  $E$  and  $r_0$  of the M1 site as a function of pressure, temperature, and composition where  $D_{\text{Ti}}$  serves also as a proxy for the prediction of  $D_{\text{Hf}}$  and  $D_{\text{Zr}}$  (cf. Blundy and Wood, 2003). Fits of predicted against measured values (Fig. 11) provide good regression statistics for  $D_{\text{Ti}}$  ( $R^2_{\text{Ti}} = 0.96$  and  $R^2_{\text{Ti}} = 0.94$ ) and  $D_{\text{Zr}}$  ( $\text{SEE}_{\text{Ti}} = 0.96$  and  $\text{SEE}_{\text{Ti}} = 0.94$ ). Considering the electrostatic effect produced by insertion of a cation of different charge from the major cation (e.g.,  $\text{Ti}^{4+}$  for  $\text{Mg}^{2+}$ ) into the crystal lattice, the proportion of M1 sites charge-balanced by a tetravalent cation decreases (from 0.018 to 0.001) with increasing  $^{M1}\text{Mg}$  from Zone 1 to Zone 2 clinopyroxenes. As demonstrated by Wood and Blundy (2001), however, cations are also likely to be present in sites of unbalanced charge. The electrostatic work done in placing HFSE into a crystal site of the wrong charge from Zone 1 to Zone 2 clinopyroxenes can be quantified by the model of Hill et al. (2011). It is found that the proportion of M1 sites charge balanced by a trivalent cation decreases (from 0.031 to 0.017) with decreasing  $^{M1}\text{Al}$ ,

confirming that the average energy of HFSE substitution is effectively related to the concentrations of the different types of charged or uncharged sites in the clinopyroxene structure (cf. Wood and Blundy, 2001). Owing to charge balancing, the decrease of  $D_{Ti}$  is also positively correlated with  $D_{REE+Y}$  (Fig. 11), as the  ${}^TAl/Si$  ratio of clinopyroxene decrease from Zone 1 to Zone 2 (Fig. 2). A similar relationship has been reported by Hack et al. (1994) and recently reappraised by global regression fits of Bedard (2014). Titanium much more readily enters the M1 site of  $CaAl_2SiO_6$  (3+ charge on M1) than  $CaMgSi_2O_6$  (2+ charge on M1). Consequently,  $D_{Ti}$  increases with increasing  ${}^TAl$  during substitution of diopside with Tschermak molecules and replacement of divalent Mg with trivalent Fe and Al (Wood and Triguila, 2001). In contrast, the substitution of REE+Y onto M2 is facilitated for the molecule  $REEMgAlSiO_6$  in which the charge on M1 does not increase with increasing  ${}^TAl$ . Indeed, M2 is almost exclusively occupied by divalent Ca, Mg, and Fe, apart from minor amounts of monovalent Na. Thus, charge imbalances associated with the entry of trivalent REE+Y cations into M2 cause  $D_{REE+Y}$  to increase with  ${}^TAl$  in concert with an increased ease of locally balancing the excess charge at M2 as the number of surrounding tetrahedral Al atoms increases (Hill et al., 2000).

When magma evolution is modelled by petrologists, the partition coefficients are frequently assumed to be constant during the entire evolution of magma owing to obvious oversimplifications that do not take into account the dependence of trace element partitioning on  $P$ ,  $T$ , and  $X$ . Undoubtedly, as the physicochemical conditions of the natural system change, a variety of partition coefficient values contribute to determining the final compositions of the solid products. To illustrate the change of REE+Y partitioning during magma differentiation, MELTS (Ghiorso and Sack, 1995) thermodynamic simulations were performed using as initial melt composition the bulk-rock analysis of the trachybasaltic lava (Table 6S). The simulations started from the melt liquidus temperature (1,170 °C) assuming pressure (100 MPa), melt- $H_2O$  content (1.5 wt.%), and oxygen fugacity (nickel-nickel oxide) comparable to those estimated for recent eruptions at Mt. Etna (Mollo et al., 2015b). The compositions of clinopyroxenes and coexisting melts recovered by MELTS code

were then used as input data for the lattice strain model of Wood and Blundy (1997).  $D_0$  values predicted for the partitioning of REE+Y during magma differentiation are plotted against  $T$  and  ${}^T\text{Al}$ , and compared with  $D_0$  measured for Zone 1 and Zone 2 clinopyroxenes (Fig. 12). Notably,  $T$  and  ${}^T\text{Al}$  influence strongly  $D_{\text{REE+Y}}$ , as well as the shape of the parabola in the Onuma diagram (Sun and Liang, 2012). Changing a single parameter at a time, it is found that  $D_0$  increases with (i) decreasing  $T$  due to the positive entropy of fusion of silicate minerals (Wood and Blundy, 1997) and (ii) increasing  ${}^T\text{Al}$  in response to local charge balance requirements (Hill et al., 2000). On this basis, the partitioning behaviour of REE+Y predicted along the evolutionary path of trachybasaltic magma is opposite to that measured during melt contamination by siliceous crustal material (Fig. 12). Over an almost identical temperature decrease of  $\sim 70$  °C, two different scenarios are observed: (1)  $D_0$  increases by  $\sim 32\%$  during magma differentiation when  ${}^T\text{Al}$  slightly decreases by  $\sim 17\%$  and (2)  $D_0$  decreases by  $\sim 19\%$  during magma contamination when  ${}^T\text{Al}$  drastically decreases by  $\sim 76\%$ . Unravelling the combined effects of  $T$  and  ${}^T\text{Al}$  is not a trivial task when the compositions of clinopyroxene and coexisting melt change systematically as a function of the intensive variables of the system. However, it is also true that much progress has been made in recent years (Yao et al., 2012; Sun and Liang, 2012, 2013; Dygert et al., 2014; Dygert and Liang, 2015; Qian et al., 2015). For example, Sun and Liang (2012) demonstrated that temperature and clinopyroxene composition may have either a competing effect or an enhancing effect on  $D_{\text{REE+Y}}$  during adiabatic mantle melting, depending on how  ${}^T\text{Al}$  in clinopyroxene varies as a function of  $T$ . Furthermore, during assimilation of olivine xenocrysts (or peridotite xenoliths) and simultaneous crystallization of clinopyroxene, Qian et al. (2015) showed that the behaviour of  $D_{\text{REE+Y}}$  and  $D_{\text{HFSE}}$  diverges with the decreasing temperature, due to inverse covariation between Al and Mg favouring Tschermak molecules. Turning to the case of magmatic evolution at Mt. Etna, the rate of temperature decrease is the dominant factor controlling  $D_0$  in scenario (1) when changes in  ${}^T\text{Al}$  are relatively small in clinopyroxenes formed during magma differentiation (Fig. 12). Conversely, in scenario (2), REE+Y become more incompatible within the clinopyroxene crystal lattice (Fig. 12), due to the marked  ${}^T\text{Al}$

deficiency that virtually overcomes the enhancing effect of temperature decrease. Extrapolation of this finding to larger spatial and temporal scales leads to the conclusion that assimilation of siliceous crustal components may favor the formation of diopsidic clinopyroxene-rich layers. However, changes of clinopyroxene chemistry due to crustal contamination have dramatic effects on the magnitude of trace element partitioning due to limited cation incorporation into the newly-formed crystals.

Following De Paolo (1981), partition coefficients for Ce and Y from scenario (1) and scenario (2) were used to model the differentiation process driven by either fractional crystallization (FC) or assimilation and fractional crystallization (AFC). These simple, but different, models may help to elucidate the control of clinopyroxene on the final trace element content of magma. Stepwise calculations were performed changing the partition coefficient at each step of fractionation as a function of temperature and clinopyroxene chemistry. Starting from the original bulk-rock composition of the trachybasaltic magma, four different degrees of clinopyroxene fractionation (i.e., 5%, 10%, 15%, and 20%) were considered over the restricted temperature range of  $\sim 70$  °C. The quartzite bulk-rock was assumed as the assimilant; the assimilation rate relative to fractional crystallization rate was set to 0.1. When both FC and AFC processes are modeled by  $D_{Ce}$  and  $D_Y$  from scenario (1), two similar trajectories are derived for the temperature-dependent partition coefficients (Fig. 13a). In contrast, Ce and Y from scenario (2) show substantial enrichments in the contaminated magma, due to the formation of diopside-rich clinopyroxenes that are highly depleted in aluminium and less inclined to incorporate REE+Y (Fig. 13a). This causes melt hybridization and subsequent clinopyroxene crystallization to produce REE+Y enrichments that cannot be modeled by simple variation of the partition coefficient along the thermal path of the system. For the sake of completeness, the clinopyroxene and coexisting melt compositions recovered by MELTS code (Table 6S) were adopted to estimate the partition coefficients of HFSE through the thermodynamic expression of Hill et al. (2011). Using the same stepwise calculations described above, the evolution of magma is modelled for  $D_{Ti}$  and  $D_{Zr}$  as a function of temperature and mineral



chemistry. Similarly to what is observed for  $D_{Ce}$  and  $D_Y$  from scenario (2), the weak incorporation of Ti and Zr into diopside-rich clinopyroxenes of hybrid origin results in an AFC trajectory much steeper than that found for the temperature-dependent partition coefficients, showing considerable HFSE enrichments in the contaminated magma relative to FC and AFC processes from scenario (1) (Fig. 13b).

## 6. Concluding remarks

The petrological consequence of Etnean magma contamination by siliceous continental crust is the formation of a hybrid melt favoring crystallization of clinopyroxenes that are compositionally distinct from those found in the uncontaminated magma. Diopside-rich clinopyroxenes from the hybrid melt result from cation exchanges in which  $M1(Al, Fe^{3+})$  substitute for  $M2(Mg, Fe^{2+})$  coupled with replacement of Si with Al in the tetrahedral site. The incorporation of trace elements in the crystal lattice is related to charge balance mechanisms primarily controlled by  $TAl$ . Specifically, hybrid clinopyroxenes depleted in Al cannot easily accommodate REE+Y in M2 as the number of local charge-balanced configurations decreases. Through the lattice strain theory, it is also found that simultaneous cation exchanges across the M1, M2, and T sites have profound effects on the partitioning of REE+Y and HFSE that, in turn, are weakly dependent on temperature and melt composition. Consequently, the behavior of REE+Y and HFSE observed during melt hybridization diverges significantly from that measured along the thermal path of the magma, leading to unexpected trace element enrichments due to low cation incorporations into the crystal lattice of hybrid clinopyroxenes.

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### Figure captions

Fig. 1. Quartzite xenolith enclosed in a thin trachybasaltic lava envelope (a). Optical photomicrograph of the trachybasaltic lava (Zone 1), the hybrid melt resulting as a boundary layer between lava and quartzite (Zone 2), and the partially melted quartzite (Zone 3) (b). SEM photomicrograph in backscattered imaging mode of Zone 1, Zone 2, and Zone 3 (c). Detailed SEM photomicrograph of Zone 2 and Zone 3 (d). Detailed SEM photomicrograph of clinopyroxenes from Zone 2 (e). Detailed SEM photomicrograph of clinopyroxenes from Zone 1 (f).

Fig. 2. Compositional variations of clinopyroxenes from Zone 1 and Zone 2 expressed in terms of cations calculated on the basis of 6 oxygen atoms per formula unit and molecules, i.e. diopside (Di), hedenbergite (Hd), enstatite (En), ferrosilite (Fs), jadeite (Jd) and tschermakitic components ( $\Sigma T$ s equals to the sum of Ca-, CaFe-, CaCr- and CaTi-Tschermak molecules).

Fig. 3. Test for equilibrium of Mollo et al. (2013b) based on the difference ( $\Delta$ ) between diopside + hedenbergite (DiHd) contents measured in the analyzed crystals with those predicted for clinopyroxene via regression analysis of clinopyroxene-melt pairs at equilibrium. Data from this study plot within 10% of the one-to-one line, suggest near-equilibrium crystallization (a). Test for equilibrium of Putirka (2008) based on the Fe–Mg cation exchange reaction [ $Kd_{Fe-Mg} = (Fe^{cpX} /$

$\text{Fe}^{\text{melt}}) \times (\text{Mg}^{\text{melt}}/\text{Mg}^{\text{cpX}})]$ . Clinopyroxene-melt pairs match with both the equilibrium ranges of  $0.27 \pm 0.03$  and  $0.28 \pm 0.08$  indicated by Putirka et al. (2003) and Putirka (2008) (b). Clinopyroxene crystallization temperatures estimated by the  $P$ - $\text{H}_2\text{O}$ -independent thermometer of Putirka et al. (1996) with uncertainty  $\pm 23$  °C (c).

Fig. 4. Compositional variations of melts from Zone 1, Zone 2, and Zone 3 corresponding to residual melt of the trachybasaltic lava, hybrid melt formed at the quartzite-lava interface, and  $\text{SiO}_2$ -rich melt resulting from partial dissolution of quartz grains respectively.

Fig. 5. Quantitative X-ray maps of the quartzite-lava interface showing progressive assimilation of the  $\text{SiO}_2$ -rich melt from quartz dissolution by the Etnean magma.

Fig. 6. Si-Al X-ray overlay map in which, among all major cations, silicon and aluminium show the greatest chemical variations.

Fig. 7. Linear least-squares mass balance calculations that couple the simple mixing process with clinopyroxene crystallization (cf. Barnes et al., 2002). The goodness of fit is corroborated by the low sums of squares of residuals.

Fig. 8. Partition coefficients from Zone 1 and Zone 2 presented as Onuma diagrams (Onuma et al., 1968) in which elements are grouped by ionic charge for M1 and M2 clinopyroxene crystallographic sites and arranged from smallest to largest ionic radius (Shannon, 1976) within each group.

Fig. 9. Partition coefficients  $D_{\text{TE}}$  (i.e.,  $D_{\text{Cr}}$  and  $D_{\text{Sc}}$ ),  $D_{\text{HFSE}}$  (i.e.,  $D_{\text{Hf}}$  and  $D_{\text{Zr}}$ ),  $D_{\text{LILE}}$  (i.e.,  $D_{\text{Pb}}$  and  $D_{\text{Sr}}$ ), and  $D_{\text{REE+Y}}$  (i.e.,  $D_{\text{La}}$  and  $D_{\text{Lu}}$ ) from Zone 1 and Zone 2 plotted against  $^{\text{T}}\text{Al}$ .

Fig. 10. Lattice strain parameters calculated by the thermodynamically-derived model of Wood and Blundy (1997) are plotted against the aluminum content within the tetrahedral site of clinopyroxene (i.e.,  ${}^T\text{Al}$ ) (a) and the number of non-bridging oxygens per tetrahedral cations in the melt (i.e., NBO/T) (b).  $D_0$  is the partition coefficient for the strain-free substitution,  $r_0$  is the radius of a hypothetical cation that substitutes into the site with zero strain, and  $E$  is the effective Young's modulus for the lattice site.

Fig. 11. Values of REE+Y (i.e.,  $D_{\text{La}}$  and  $D_{\text{Lu}}$ ) and HFSE (i.e.,  $D_{\text{Ti}}$  and  $D_{\text{Zr}}$ ) partition coefficients measured from this study are compared with those predicted by the models of Wood and Blundy (1997) and Hill et al. (2011), respectively. Linear regression fits of data yield high correlation coefficients ( $R^2$ ) and low standard errors of estimate (SEE).

Fig. 12.  $D_0$  values predicted for the partitioning of REE+Y during magma differentiation are plotted against  $T$  and  ${}^T\text{Al}$ , and compared with the strain-free partition coefficients measured for Zone 1 and Zone 2 clinopyroxenes.

Fig. 13. Ce and Y (a), and Ti and Zr (b) partition coefficients as a function of temperature and clinopyroxene composition are used to illustrate magma dynamics driven by fractional crystallization (FC) and assimilation and fractional crystallization (AFC) processes.



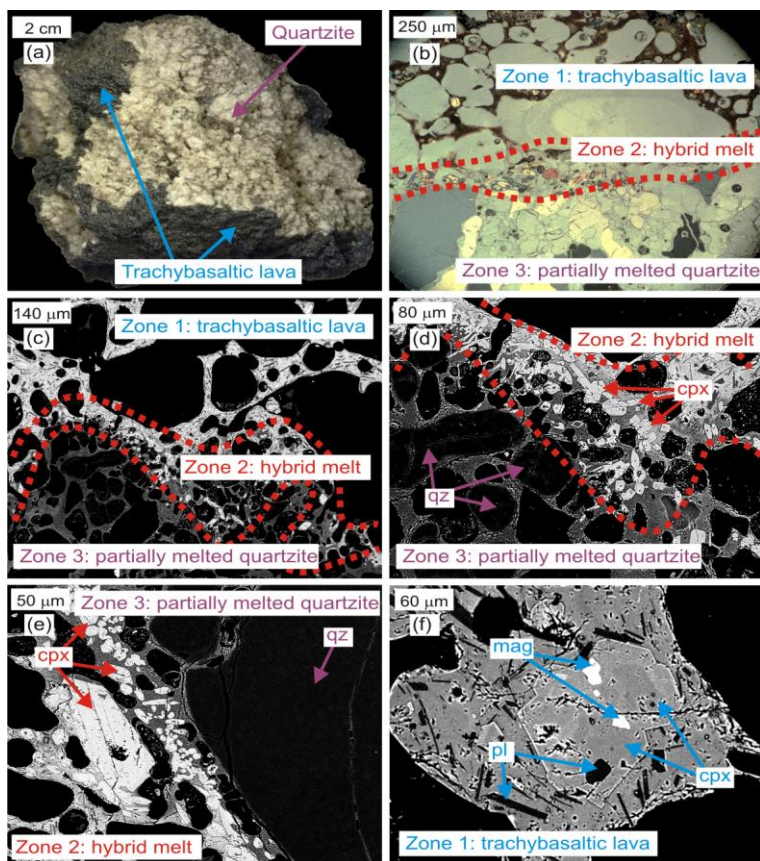


Figure 1

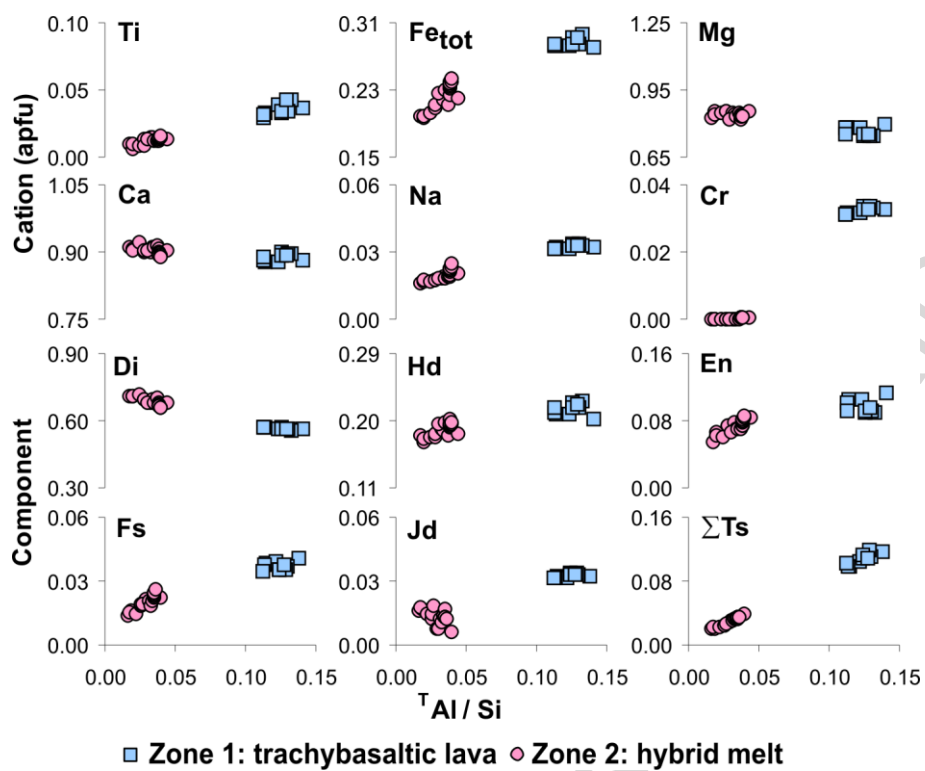


Figure 2

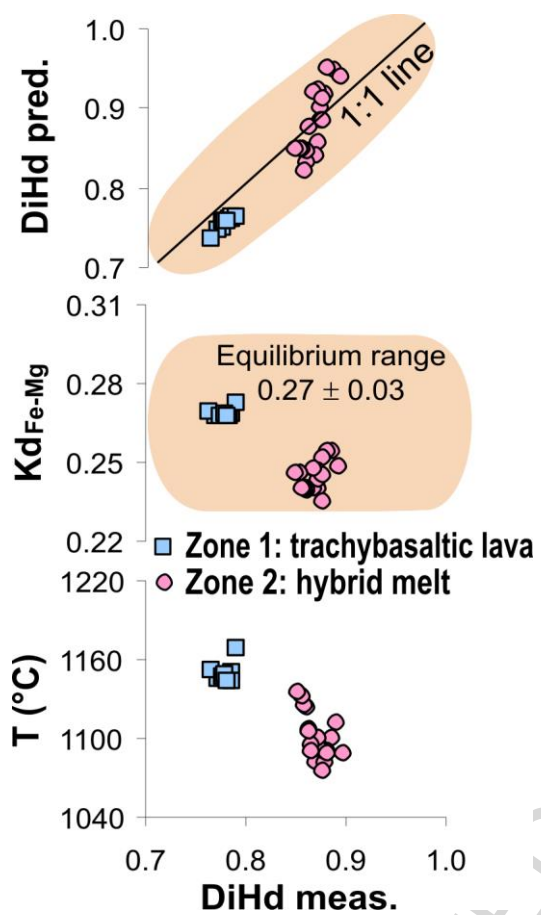


Figure 3

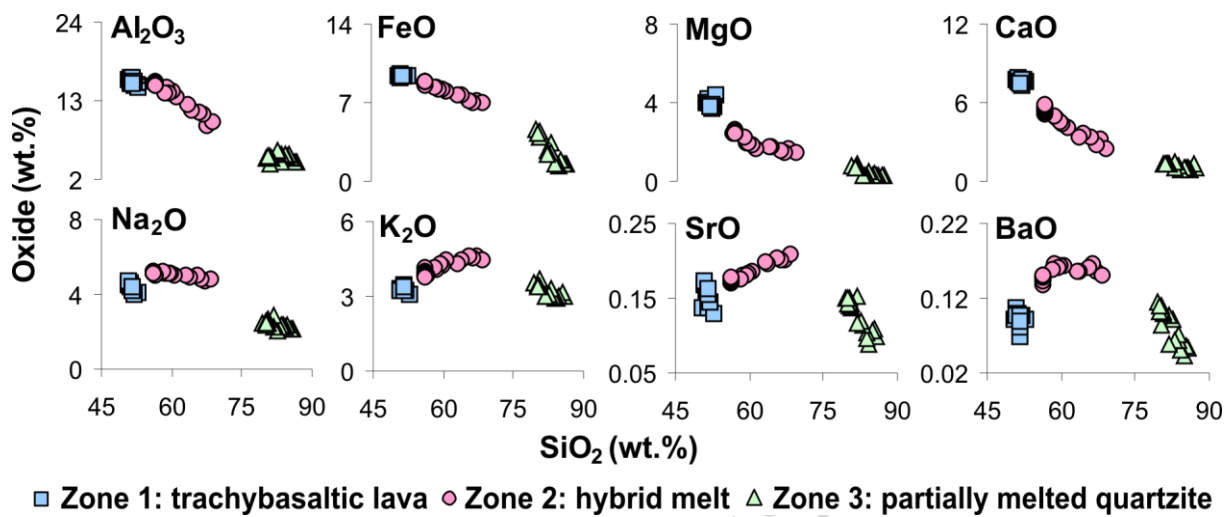


Figure 4

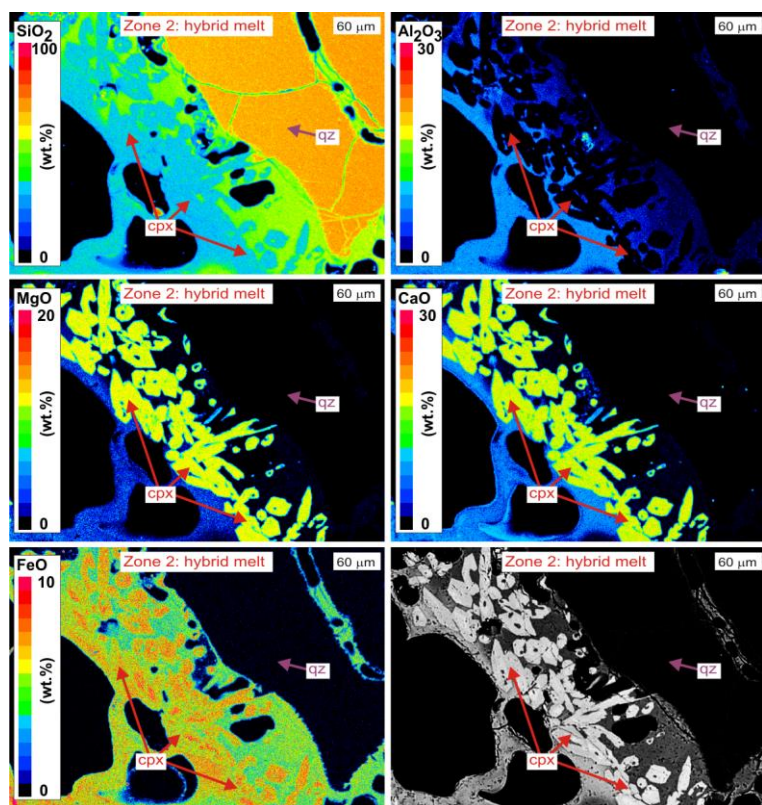


Figure 5

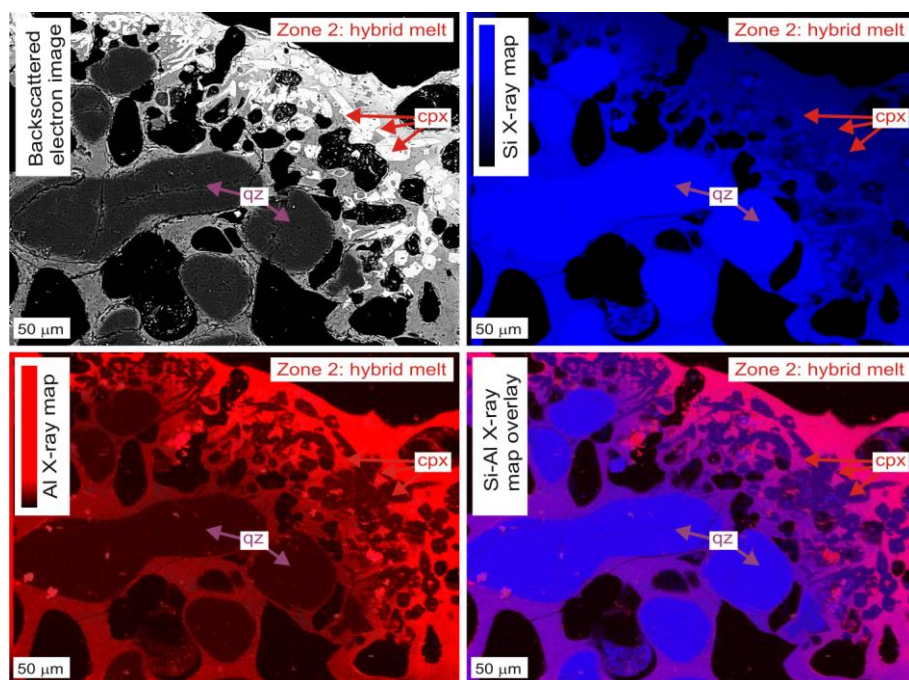


Figure 6

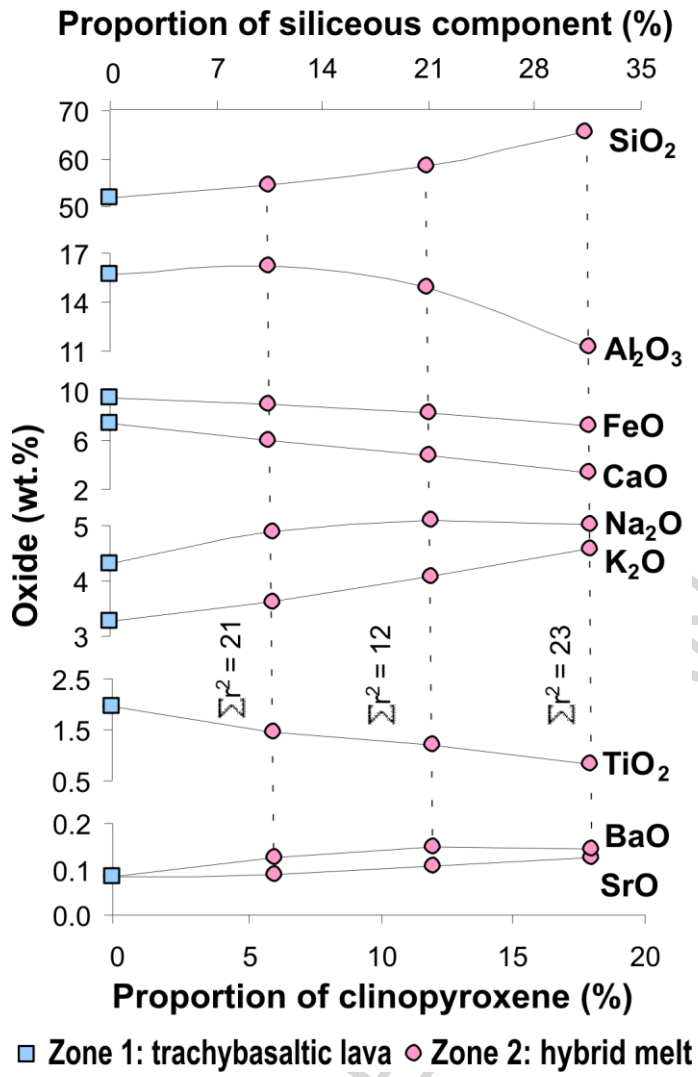


Figure 7

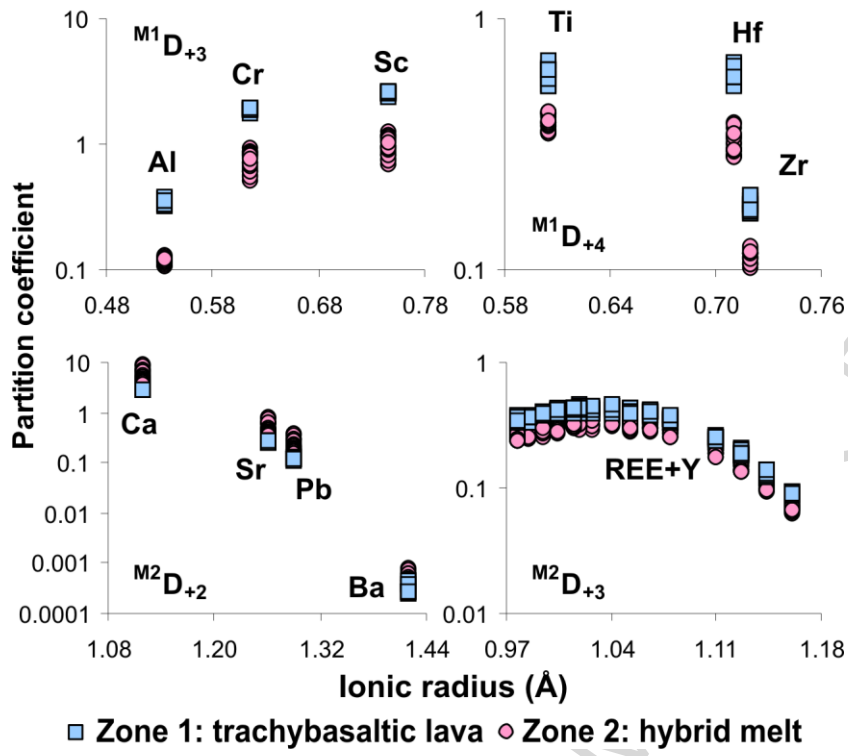


Figure 8



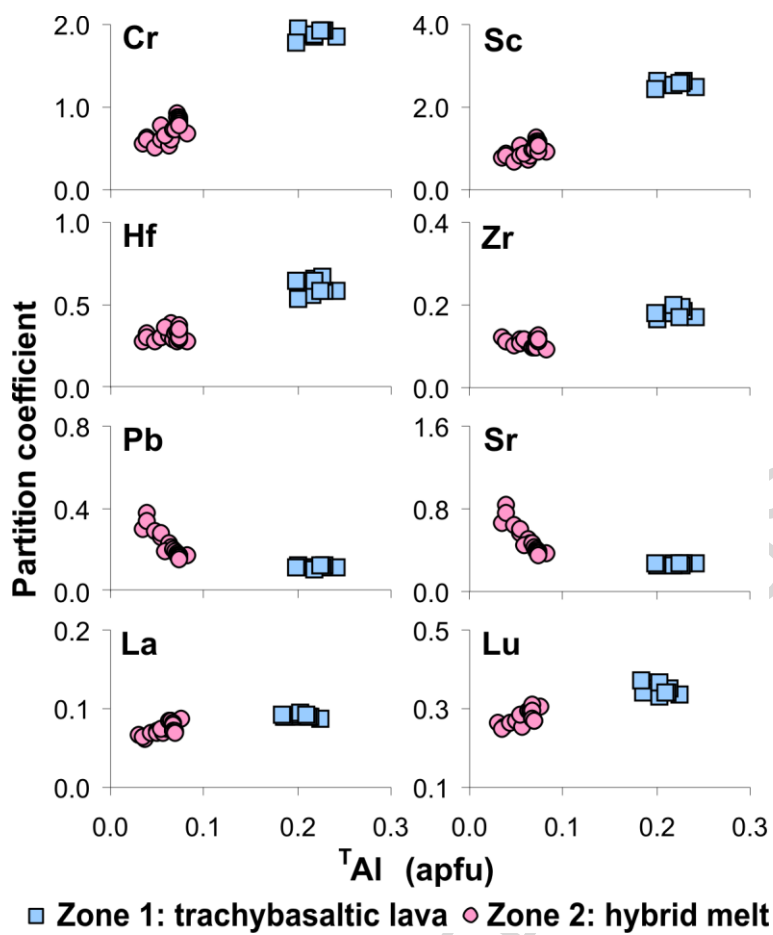


Figure 9

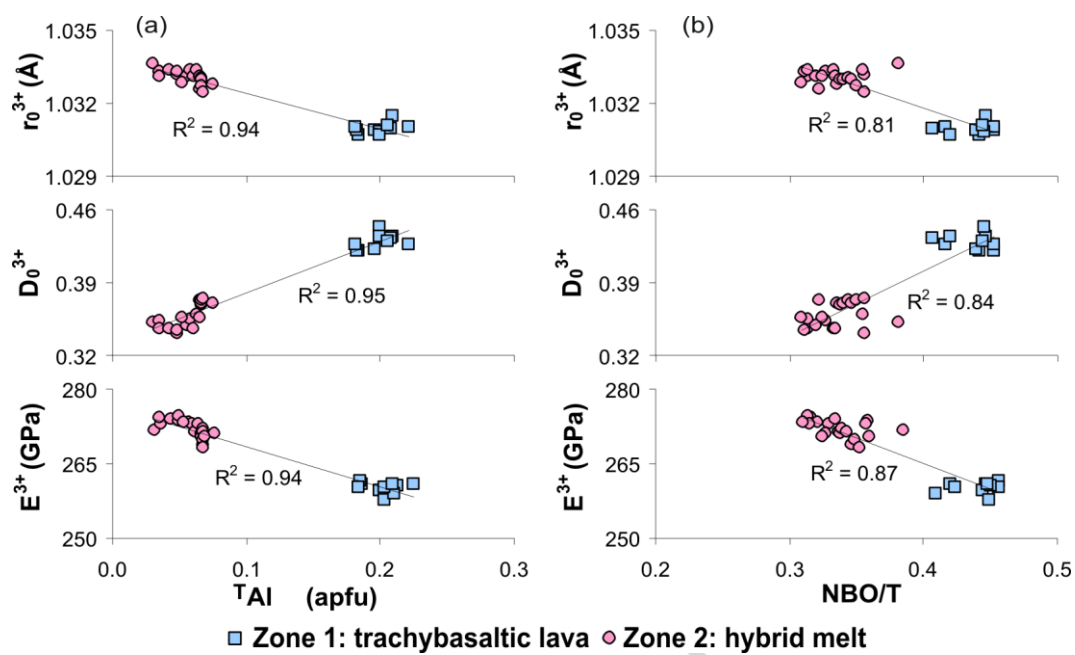


Figure 10

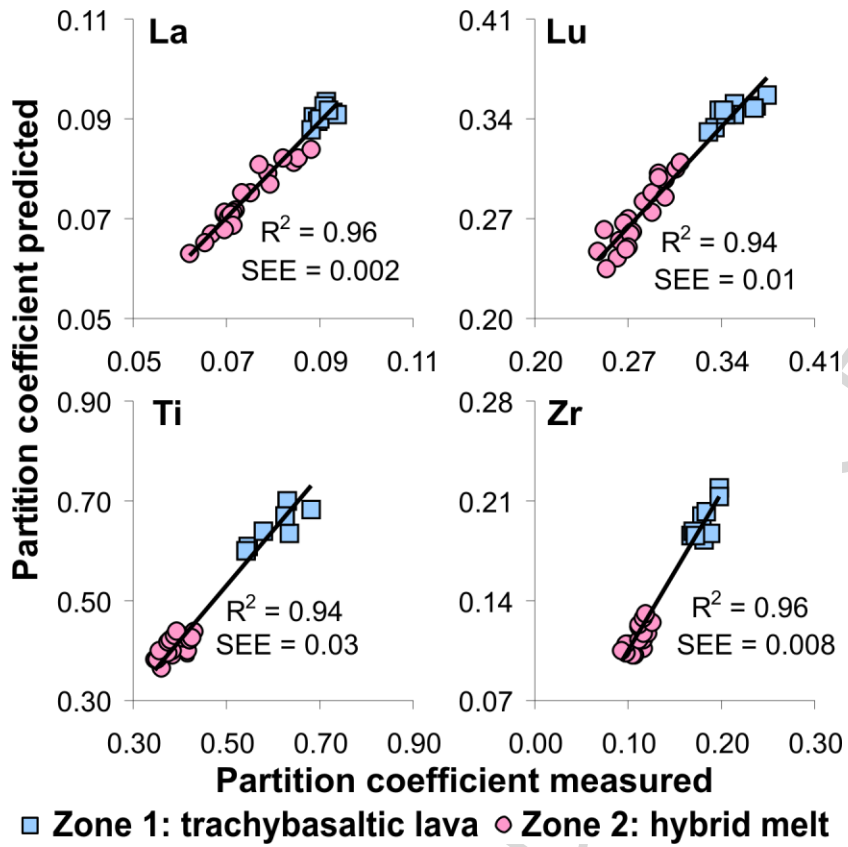


Figure 11

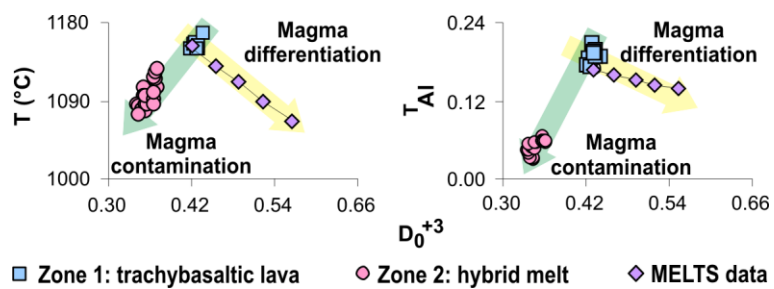


Figure 12

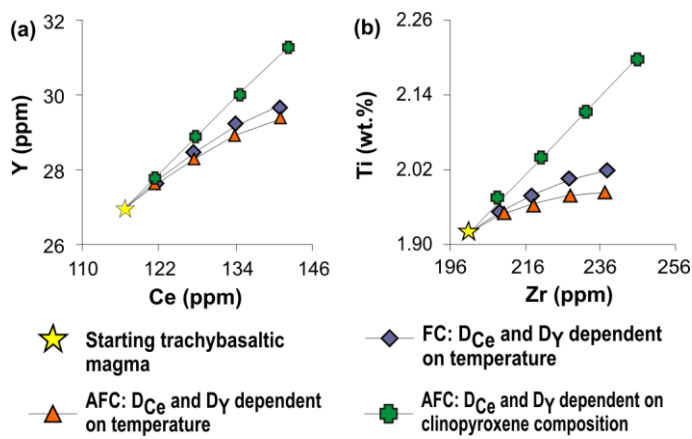


Figure 13

**Research Highlights**

Siliceous crustal xenoliths are sometimes erupted at Mt. Etna volcano > Magma contamination is accompanied by hybrid melts and newly-formed clinopyroxenes > Cation exchanges across the M1, M2, and T sites of clinopyroxene occur > Partition coefficients diverge from those derived by magmatic differentiation.